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9-ACETOXYANTHRACENE DERIVATIVES. PART VII. SYNTHESSES AND ELUCIDATION OF THE MOLECULAR STRUCTURE OF 9-ACETOXY-10-ARYLANTHRACENES FROM FLUOROPHENOLS

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SUMMARY

The syntheses and elucidation of the molecular structures of the new 9-acetoxy-10-arylanthracenes from o-, m- and p-fluorophenols are described.

INTRODUCTION

In the course of our investigations [1] on 9-acetoxyanthracene derivatives, we have described the syntheses and elucidated the molecular structure of three new 9-acetoxy-10-arylanthracenes from o-, m- and p-fluorophenols. These compounds probably can be used in dye lasers for generating electromagnetic waves in the range of 360-410 nm [2].

RESULTS AND DISCUSSION

9-Acetoxy-10-arylanthracenes from o-, m- and p-fluorophenols have been obtained as a result of multi-step syntheses of the new compounds.

Condensation of 2-benzoylbenzoic acid with o-fluorophenol gave 3-phenyl-3-(3'-fluoro-4'-hydroxyphenyl) phthalide (1): with m-fluorophenol, 3-phenyl-3-(2'-fluoro-4'-hydroxyphenyl)-phthalide (2): and with p-fluorophenol, 3-phenyl-3-(5'-fluoro-2'-hydroxyphenyl) phthalide (3). Acetylation of phthalides

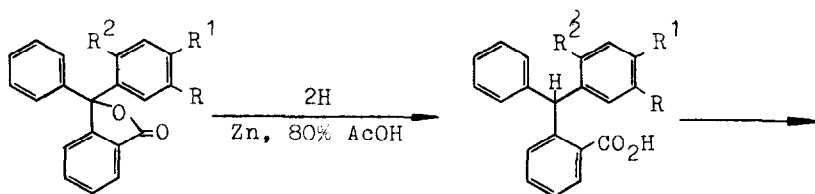
1-3 with  $\text{Ac}_2\text{O}/\text{AcONa}$  gave their acetates 1a-3a. After reduction of compounds 1-3 with Zn-dust in 80% AcOH the corresponding derivatives of 2-benzhydrylbenzoic acid (1b-3b) have been obtained.

During acetylation of acids 1b-3b with  $\text{Ac}_2\text{O}$  in the presence of traces concentrated  $\text{H}_2\text{SO}_4$  there takes place also intracyclization to:

9-acetoxy-10-(4'-acetoxy-3'-fluorophenyl)anthracene (1c),

9-acetoxy-10-(4'-acetoxy-2'-fluorophenyl)anthracene (2c) and

9-acetoxy-10-(2'-acetoxy-5'-fluorophenyl)anthracene (3c).



1: R=F,  $\text{R}^1=\text{OH}$ ,  $\text{R}^2=\text{H}$ ;

2: R=H,  $\text{R}^1=\text{OH}$ ,  $\text{R}^2=\text{F}$ ;

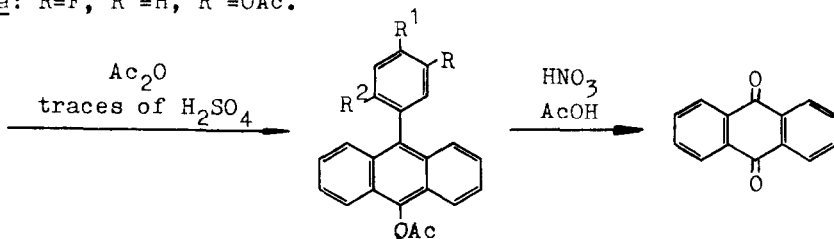
3: R=F,  $\text{R}^1=\text{H}$ ,  $\text{R}^2=\text{OH}$ ;

1b }  
2b } substituents  
3b } as in 1-3

1a: R=F,  $\text{R}^1=\text{OAc}$ ,  $\text{R}^2=\text{H}$ ;

2a: R=H,  $\text{R}^1=\text{OAc}$ ,  $\text{R}^2=\text{F}$ ;

3a: R=F,  $\text{R}^1=\text{H}$ ,  $\text{R}^2=\text{OAc}$ .



1c }  
2c } substituents  
3c } as in 1a-3a

4c: R= $\text{R}^1=\text{R}^2=\text{H}$ ;

5c: R= $\text{R}^1=\text{H}$ ,  $\text{R}^2=\text{OAc}$ ;

6c: R= $\text{R}^2=\text{H}$ ,  $\text{R}^1=\text{OAc}$ .

The structures of compounds 1c-3c have been confirmed with the use of chemical and spectral PMR methods [3,4]. Boiling of compounds 1c-3c with  $\text{HNO}_3$  in AcOH according to the method [3] caused their degradation to anthraquinone, which confirmed that substituents are in the phenyl ring and not in the anthracene skeleton.

The comparison of PMR spectra of 1c-3c with those of 9-acetoxy-10-phenylanthracene (4c), 9-acetoxy-10-(2'-acetoxyphenyl)anthracene (5c) and 9-acetoxy-10-(4'-acetoxyphenyl)-anthracene (6c) elucidated the exact positions of substituents in phenyl ring according to a given Scheme. For compound 4c : 9-OAc  $\delta$  = 2.65 ppm; 5c: 9-OAc  $\delta$  = 2.51, 2'-OAc  $\delta$  = 1.35 ppm; 6c: 9-OAc  $\delta$  = 2.60, 4'-OAc  $\delta$  = 2.33 ppm [4].

The elucidation of the molecular structures of 1c-3c indicates simultaneously also the structures of the initial compounds 1-3. This is very important especially in the case of m-fluorophenol, because in m-halogenophenols there is also very active hydrogen atom in o-position to OH group [5].

#### EXPERIMENTAL

All m.p.s reported below are corrected. Infrared (IR) spectra were measured in Nujol (within the ranges 650-1300 and 1500-2000  $\text{cm}^{-1}$ ) and hexachlorobutadiene (within the ranges 1300-1500 and 2000-3600  $\text{cm}^{-1}$ ) mulls on a Unicam SP-200 spectrophotometer. The  $\nu_{\text{max}}$  are reported only for characteristic groups. Proton magnetic resonance (PMR) spectra were run with a Jeol LNM-4H-100 (80 MHz) spectrometer in deuteriochloroform solution; as a internal standard tetramethylsilane was used (TMS,  $\delta$  = 0.00 ppm).

#### 3-Phenyl-3-(fluorohydroxyphenyl) phthalides (1-3)(nc)

To a mixture of 0.01 mole of 2-benzoylbenzoic acid and 0.011 mole of the appropriate fluorophenol, 3 g of anhydrous  $\text{ZnCl}_2$  was added. The reaction mixture was heated then on an oil bath for ca 8 hrs at 120-125°C. The melt was dissolved in hot AcOH and solution poured into water. The white amorphous precipitate was filtered off, washed well with water, extracted

with 5% NaOH aq., filtered off again and washed several times with water. The precipitate (yield of crude compounds ca. 80-85%) after several recrystallizations from EtOH yielded crystals (for results see Table).

### 3-Phenyl-3-(acetoxyfluorophenyl)phthalides (1a-3a) (nc)

A sample of 0.5 g of phthalide (1-3) was treated with 5 ccm of Ac<sub>2</sub>O and ca. 0.5 g of anhydrous AcONa and heated for 3 hrs on the oil bath at 120°C. Then the mixture was poured into water. The precipitate was filtered off, washed with water (yield ca. 80-85%) and recrystallized several times from EtOH. For results see Table.

### 2-Fluorohydroxybenzhydrylbenzoic acids (1b-3b) (nc)

A sample of 5.00 g of phthalide (1-3) was reduced with 10 g of Zn-dust in a solution of 150 ccm of 80% AcOH during 7-10 hrs heating under a reflux condenser. Then the solution was decanted into 800 ccm of water. The white amorphous precipitate was filtered off (yield ca. 90%) and recrystallized from AcOH or EtOH (for results see Table).

### 9-Acetoxy-10-(acetoxyfluorophenyl)anthracene (1c-3c) (nc)

By heating of 1 g of acid (1b-3b) in 5 ccm of acetic anhydride and 0.03 ccm of concd. sulfuric acid for 1 hr under reflux, pouring the solution into 500 ccm of water and recrystallizing the crude product (yield ca. 85%) from EtOH, compounds 1c-3c were obtained (for results cf. Table and Fig.). Solutions of 1c-3c in organic solvents exhibited blue fluorescence.

Compounds 1c-3c after heating in HNO<sub>3</sub>/AcOH according to the method [3] gave anthraquinone in each case.

#### ACKNOWLEDGEMENT

Thanks are due to Mrs. T. Rakowska for her syntheses of two phthalides in experimental part.

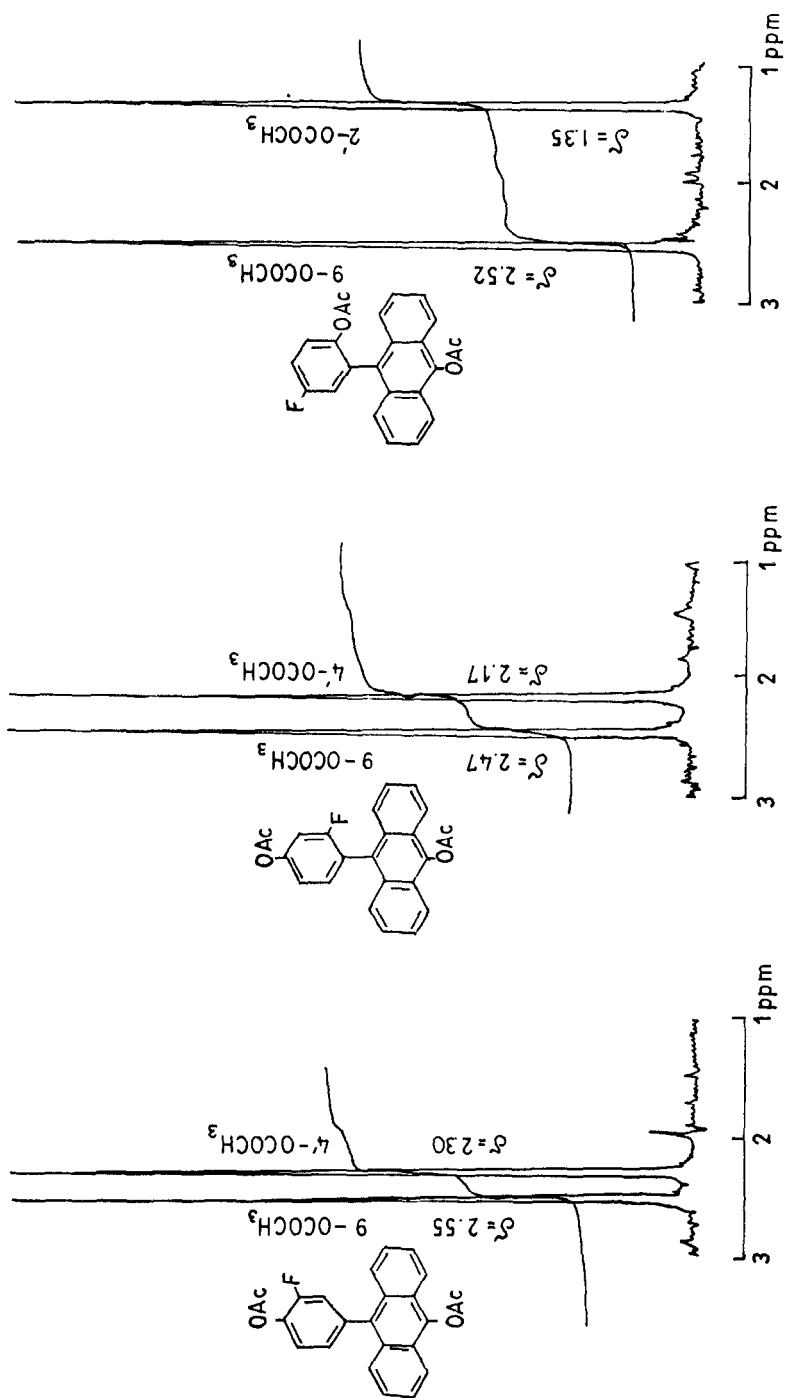


Fig. PMR spectra of 9-acetoxy-10-arylanthracenes

TABLE

Results and elemental analyses, m.p.s and IR spectra of 3-phenyl-3-(fluorohydroxyphenyl)-phthalides (1-3), their acetates (1a-3a), 2-(fluorohydroxybenzhydril)benzoic acids (1b-3b) and 9-acetoxy-10-(acetoxyfluorophenyl)anthracene (1c-3c)

No	Compound	Formula, molecular mass	Analyses [%]		M.p. [°C] EtOH	IR spectra, cm <sup>-1</sup>
			Calc.	Found		
<u>1</u>	3-Phenyl-3-(3'-fluoro-4'-hydroxyphenyl)-phthalide	C <sub>20</sub> H <sub>13</sub> O <sub>3</sub> F 320.3	C 75.0 H 4.1 F 5.9	C 75.3 H 4.2 F 5.8	200.5-202.5	1720s (C=O, γ-lactone), 1340m (C-O, phenol), 980m (C-O-C, γ-lactone);
<u>2</u>	3-Phenyl-3-(2'-fluoro-4'-hydroxyphenyl)-phthalide	C <sub>20</sub> H <sub>13</sub> O <sub>3</sub> F 320.3	-	-	-	white, amorphous substance, but crystalline are its derivatives: <u>2a</u> , <u>2b</u> and <u>2c</u>
<u>3</u>	3-Phenyl-3-(5'-fluoro-2'-hydroxyphenyl)-phthalide	C <sub>20</sub> H <sub>13</sub> O <sub>3</sub> F 320.3	C 75.0 H 4.1 F 5.9	C 75.2 H 4.0 F 6.2	252.4-253.8	1730s (C=O, γ-lactone), 1340m (C-O, phenol), 980m (C-O-C, γ-lactone);
<u>1a</u>	Acetate of phthalide <u>1</u>	C <sub>22</sub> H <sub>15</sub> O <sub>4</sub> F 362.4	C 72.9 H 4.2 F 5.2	C 72.5 H 4.1 F 5.5	132.0-133.8	1765s (C=O of COOCH <sub>3</sub> ), 1380s (CH <sub>3</sub> , acetate), 1205s (C-O-C, acetate);
<u>2a</u>	Acetate of phthalide <u>2</u>	C <sub>22</sub> H <sub>15</sub> O <sub>4</sub> F 362.4	C 72.9 H 4.2 F 5.2	C 72.7 H 4.0 F 5.7	132.5-134.1	1750s (C=O of COOCH <sub>3</sub> ), 1390s (CH <sub>3</sub> , acetate), 1180s (C-O-C, acetate);
<u>3a</u>	Acetate of phthalide <u>3</u>	C <sub>22</sub> H <sub>15</sub> O <sub>4</sub> F 362.4	C 72.9 H 4.2 F 5.2	C 72.6 H 4.0 F 5.3	122.3-124.0	1750s (C=O of COOCH <sub>3</sub> ), 1380s (CH <sub>3</sub> , acetate), 1200s (C-O-C, acetate);

<u>1b</u>	2-(3',-Fluoro-4',-hydroxy- benzhydryl) benzoic acid	C <sub>20</sub> H <sub>15</sub> O <sub>3</sub> F 322.3	C 74.5 H 4.7 F 5.9	C 74.7 H 4.8 F 5.7	199.7-201.2	2770w (OH, dimer COOH), 1700s (C=O, COOH), 1290s (C-O, phenol);
<u>2b</u>	2-(2',-Fluoro-4',-hydroxy- benzhydryl) benzoic acid	C <sub>20</sub> H <sub>15</sub> O <sub>3</sub> F 322.3	C 74.5 H 4.7 F 5.9	C 74.8 H 5.1 F 5.2	219.6-221.4	2650w (OH, dimer COOH), 1720s (C=O, COOH), 1260s (C-O, phenol);
<u>3b</u>	2-(5',-Fluoro-2',-hydroxy- benzhydryl) benzoic acid	C <sub>20</sub> H <sub>15</sub> O <sub>3</sub> F 322.3	C 74.5 H 4.7 F 5.9	C 74.6 H 4.7 F 6.1	175.7-177.1	2830w (OH, dimer COOH), 1690s (C=O, COOH), 1270s (C-O, phenol);
<u>1c</u>	9-Acetoxy-10-(4'-acetoxy- 3',-fluorophenyl)- anthracene	C <sub>24</sub> H <sub>17</sub> O <sub>4</sub> F 388.4	C 74.2 H 4.4 F 4.9	C 74.3 H 4.4	240.2-241.7	1760sb (C=O of COOCH <sub>3</sub> ), 1375s (CH <sub>2</sub> , acetate), 1200s (C-O-C, acetate);
<u>2c</u>	9-Acetoxy-10-(4'-acetoxy- 2',-fluorophenyl)- anthracene	C <sub>24</sub> H <sub>17</sub> O <sub>4</sub> F 388.4	C 74.2 H 4.4 F 4.9	C 73.9 H 4.6	154.6-155.8	1770sb (C=O of COOCH <sub>3</sub> ), 1375s (CH <sub>2</sub> , acetate), 1175s (C-O-C, acetate);
<u>3c</u>	9-Acetoxy-10-(2'-acetoxy- 5',-fluorophenyl)- anthracene	C <sub>24</sub> H <sub>17</sub> O <sub>4</sub> F 388.4	C 74.2 H 4.4 F 4.9	C 74.0 H 4.5	190.4-192.1	1765sb (C=O of COOCH <sub>3</sub> ), 1375s (CH <sub>2</sub> , acetate), 1205sb (C-O-C, acetate).

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